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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.003 Å R factor = 0.042 wR factor = 0.114 Data-to-parameter ratio = 11.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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The dihydropyran ring in the title compound, $C_{19}H_{22}O_3$, adopts a half-chair conformation and the cyclohexane ring is in a chair conformation. In addition to van der Waals

interactions, the molecular packing in the crystal is stabilized

by $C-H \cdots \pi$ interactions, involving the dihydropyran ring of

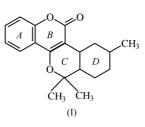
1,2,3,4,4a,12a-Hexahydro-2,5,5-trimethyl-

1H-[2]benzopyrano[3,2-c]coumarin

Comment

the symmetry-related molecules.

Coumarin derivatives possess a variety of biological activities (Crombic *et al.*, 1985). They are used in oral anticoagulation therapy (Cole *et al.*, 1988; Greenfield, 1988) and in enzyme inhibitors (Michel & Durant, 1976). Coumarin derivatives exhibit solid-state photochemical reactions (Gnanaguru *et al.*, 1985). Some amino- and hydroxycoumarin derivatives are used as laser dyes (Masilamani, 1979). As part of our studies on these derivatives, the X-ray structure determination of the title compound, (I), was undertaken.



The title molecule (Fig. 1) contains a fused four-ring system A/B/C/D. The geometry of the coumarin ring system (Table 1) is comparable to that observed in other coumarin derivatives (Chinnakali et al., 1998, 1999; Krishna et al., 2003). The bond lengths in the dihydropyran ring (C) and cyclohexane ring (D)have normal values (Allen et al., 1987). In the coumarin moiety, the pyran ring is planar within 0.056 (2) Å and the dihedral angle between the weighted least-squares planes through the benzene and pyran rings (A and B) is $3.55 (6)^{\circ}$. The dihydropyran ring adopts a half-chair conformation with asymmetry parameter $\Delta C_2(C12 - C17)$ equal to 0.004 (1) (Nardelli, 1983). The cyclohexane ring (D) adopts a chair conformation and the methyl group is equatorially attached to it. The C/D ring junction is *trans*. An intramolecular C-H···O hydrogen bond involving the carbonyl O atom is observed. The dihydropyran ring of the symmetryrelated molecule at $(\frac{1}{2} + x, \frac{3}{2} - y, -z)$ is involved in a C-H··· π interaction. In Table 2, Cg denotes the centroid of the pyran ring.

Experimental

To a refluxing solution of 4-hydroxycoumarin (1 mmol) in 10 ml of dry ethanol, citranellal (1 mmol) was added and the reaction mixture

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was refluxed for 4 h. Evaporation of the solvent in a rotory evaporator, followed by flash silica-gel chromatography, yielded the title compound. The compound was recrystallized from methanol/ chloroform (1:1) by slow evaporation.

Mo $K\alpha$ radiation

reflections

 $\theta = 2.9-27.6^{\circ}$ $\mu = 0.08 \text{ mm}^{-1}$

T = 293 (2) K

Block, colourless $0.50 \times 0.16 \times 0.14 \text{ mm}$

Cell parameters from 6654

 $w = 1/[\sigma^2(F_o^2) + (0.0647P)^2]$

where $P = (F_o^2 + 2F_c^2)/3$

+ 0.1224P]

 $(\Delta/\sigma)_{\rm max} < 0.001$

 $\Delta \rho_{\rm max} = 0.21 \text{ e} \,\bar{\text{\AA}}^{-3}$

 $\Delta \rho_{\rm min} = -0.21 \text{ e } \text{\AA}^{-3}$

Crystal data

C₁₉H₂₂O₃ $M_r = 298.37$ Orthorhombic, $P2_12_12_1$ a = 8.8291 (4) Å b = 9.8819 (4) Å c = 18.2053 (8) Å V = 1588.38 (12) Å³ Z = 4 $D_x = 1.248$ Mg m⁻³

Data collection

Siemens SMART CCD areadetector diffractometer1960 reflections with $I > 2\sigma(I)$ ω scans $\theta_{max} = 0.034$ ω scans $\theta_{max} = 28.3^{\circ}$ Absorption correction: none $h = -11 \rightarrow 11$ 11019 measured reflections $k = -12 \rightarrow 13$ 2247 independent reflections $l = -22 \rightarrow 24$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.042$ $wR(F^2) = 0.114$ S = 1.092247 reflections 199 parameters H-atom parameters constrained

Table 1

Selected geometric parameters (Å, °).

O1-C9	1.375 (2)	C5-C6	1.377 (3)
O1-C2	1.392 (2)	C5-C10	1.397 (3)
C2-O22	1.200 (2)	C6-C7	1.387 (3)
C2-C3	1.454 (3)	C7-C8	1.380 (3)
C3-C4	1.358 (3)	C8-C9	1.385 (3)
C3-C18	1.512 (3)	C9-C10	1.391 (3)
C4-O11	1.340 (2)	O11-C12	1.474 (2)
C4-C10	1.452 (3)		
O22-C2-O1	115.51 (17)	O1-C9-C8	117.71 (17)
022-C2-C3	126.39 (19)	C5-C10-C4	123.80 (17)

Table 2

Hydrogen-bonding geometry (Å, °).

$D - H \cdot \cdot \cdot A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
C16-H16B···O22	0.97	2.42	3.061 (3)	123
C19-H19 $A \cdots Cg^{i}$	0.96	2.90	3.681 (3)	140

Symmetry code: (i) $\frac{1}{2} + x, \frac{3}{2} - y, -z$.

H atoms were positioned geometrically and were treated as riding on their parent C atoms, with aromatic C-H distances of 0.93 Å, methyl C-H distances of 0.96 Å, methylene C-H distances of 0.98 Å and ethylene C-H distances of 0.97 Å. Due to the lack of anomalous scatterers, the absolute configuration was not determined

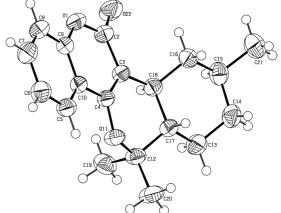


Figure 1

The molecular structure of the title compound, showing 50% probability displacement ellipsoids and the atom-numbering scheme.

from the X-ray diffraction data and the Friedel pairs were merged. The absolute configuration is unknown.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *ORTEP*-3 (Farrugia, 1997) and *PLATON* (Spek, 1990); software used to prepare material for publication: *SHELXL*97 and *PARST* (Nardelli, 1995).

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